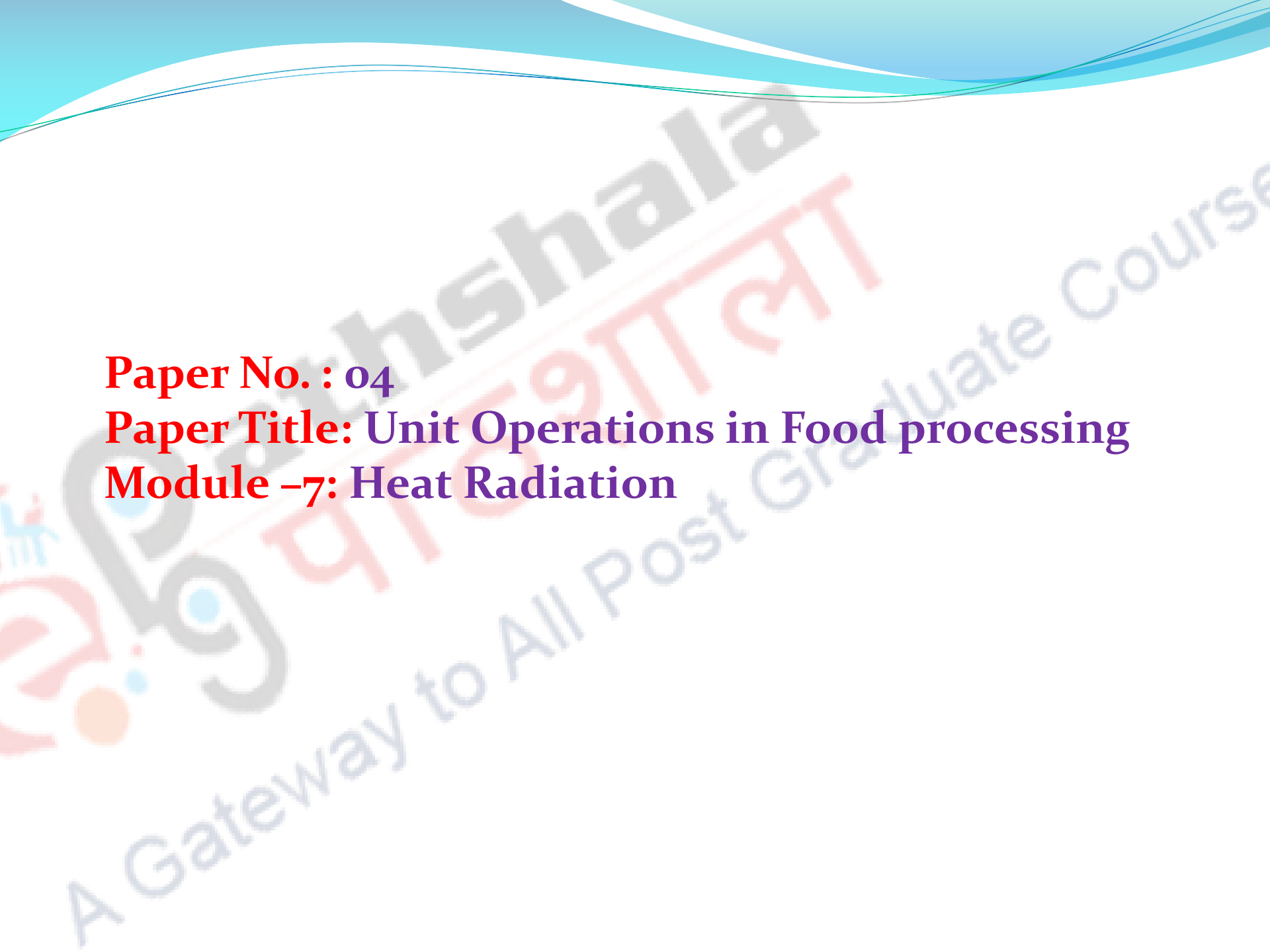


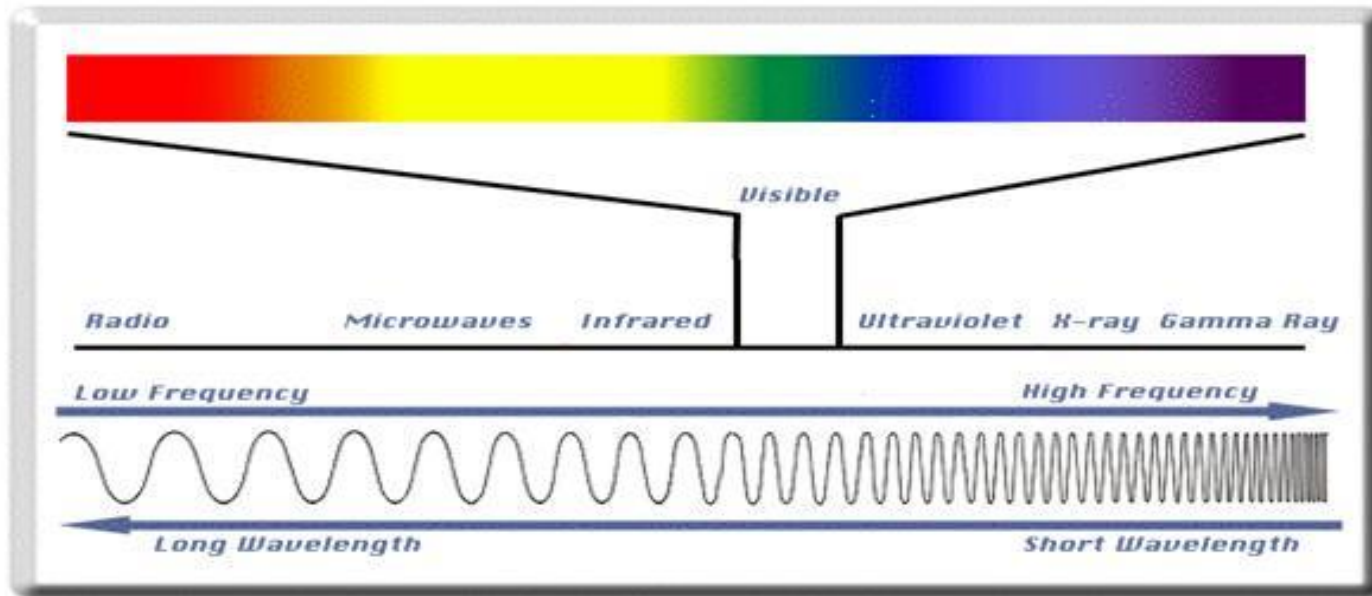
Paper No. : 04

Paper Title: Unit Operations in Food processing

Module -7: Heat Radiation



Introduction



- Emission of energy in the form of electromagnetic waves
- Depends on body temperature
- Can travel in complete vacuum
- Rate of heat transfer is faster
- Example: Baking, sunshine, Boiler

Surface characteristics

Absorptivity,

$$\alpha = \frac{q_{abs}}{q_{inc}}$$

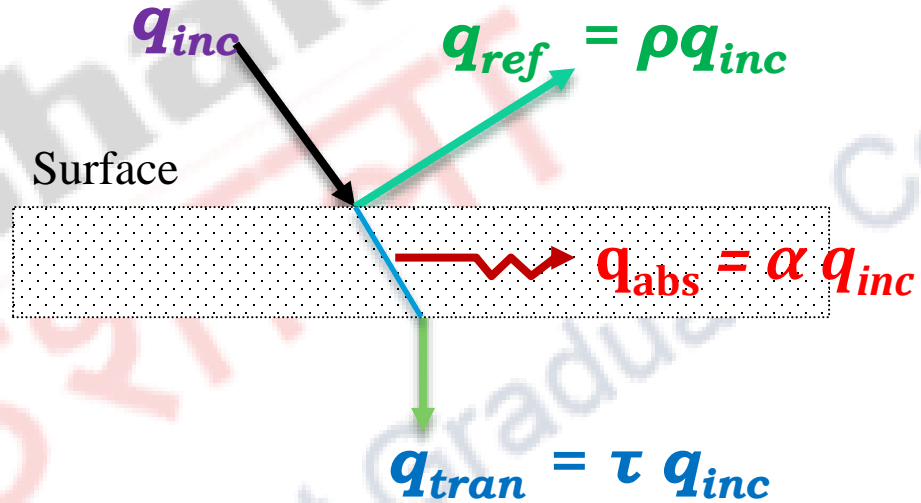
Reflectivity,

$$\rho = \frac{q_{ref}}{q_{inc}}$$

Transmissivity,

$$\tau = \frac{q_{tran}}{q_{inc}}$$

$$\alpha + \rho + \tau = 1$$



- The radiation energy is presented as energy flux (W/m^2)
- Polished surface reflects all the incident radiation $\rho = 1$
- Black body absorbs all the incident radiation, So, $\alpha = 1$

Types of Body

Black Body

A body is said to be perfectly black if it absorbs all the incident radiation to it.

$$\text{absorptivity, } \alpha = 1$$

$$\text{emissivity, } \varepsilon = 1$$

$$\text{So, } \alpha = \varepsilon = 1$$

Gray Body

A body is said to be gray if the monochromatic properties are constant at all wavelengths.

$$\text{absorptivity, } \alpha < 1$$

$$\text{emissivity, } \varepsilon < 1$$

Opaque Body

A body is said to be opaque if no incident radiation gets transmitted through the body.

$$\alpha + \rho = 1$$

Emissivity

- Emissivity is the ratio of radiant heat flux of a real body to that of a perfect black body
- Black body is a perfect emitter
- Energy emitted by a real surface is always less than energy emitted by a black surface at same temperature

Emissivity, $\varepsilon = \frac{E}{E_0}$

Radiation of a body

Radiation of a blackbody

Kirchhoff's Law

Emissivity, $\varepsilon = \frac{E}{E_0}$

Absorptivity, $\alpha = \frac{E}{E_0}$

At thermal equilibrium, $\varepsilon = \alpha$

So, Kirchhoff's law states that a body at constant temperature which is in equilibrium with its surroundings, the amount of energy absorbed by radiation is exactly same as that emitted.

Stefan-Boltzmann Law

It states that the total emissive power is the total radiation energy per unit area leaving a surface with temperature T over all wavelengths.

The diagram shows the equation $\frac{q}{A} = \sigma \epsilon T^4$ with arrows pointing from labels to the variables: $\frac{q}{A}$ is labeled 'Radiation power (Watt)', A is labeled 'Surface Area (m²)', σ is labeled 'Stefan-Boltzmann constant (5.669 X 10⁻⁸ W/m²K⁴)', ϵ is labeled 'emissivity', and T is labeled 'Absolute temperature (K)'.

$$\frac{q}{A} = \sigma \epsilon T^4$$

Radiation power (Watt)

emissivity

Absolute temperature (K)

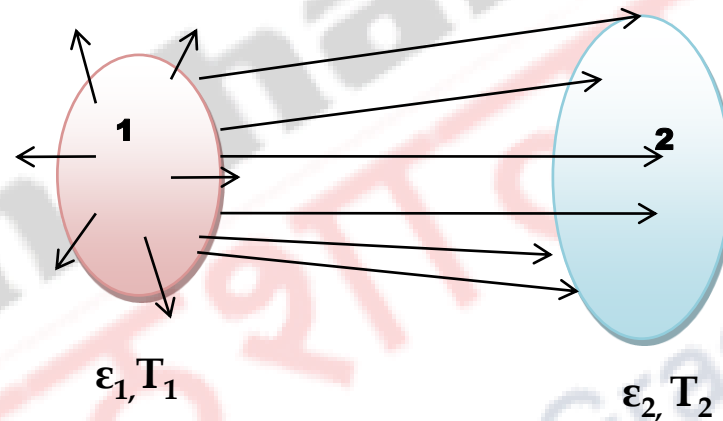
Surface Area (m²)

Stefan-Boltzmann constant (5.669 X 10⁻⁸ W/m²K⁴)

For a perfectly black body,

$$\frac{q}{A} = \sigma T^4$$

Radiation between two surfaces



$$q_{1 \rightarrow 2} = \left[\begin{array}{l} \text{radiation leaving surface 1} \\ \text{that strikes surface 2} \end{array} \right] - \left[\begin{array}{l} \text{radiation leaving surface 2} \\ \text{that strikes surface 1} \end{array} \right]$$

$$q_{12} = \sigma A K \left[T_1^4 - T_2^4 \right]$$

$$\frac{1}{K} = \frac{1}{\epsilon_1} + \frac{1}{\epsilon_2} - 1$$

Radiation from the surrounding

Example: Heat transfer from the surrounding to the bread loaf inside a baking oven

$$q_r = h_r A (T_1 - T_2)$$

$$q_{12} = \sigma A (T_1^4 - T_2^4)$$

$$h_r = \sigma \frac{(T_1^4 - T_2^4)}{T_1 - T_2}$$

$$q_{total} = q_r + q_c$$

Radiating heat transfer medium

1. Gases with high concentration
2. Asymmetric molecules like H_2O , CO_2 , CO , SO_2 etc.
3. Absorption and emission depends on the gas concentration

THANK YOU

